

# Nuclear Data and Society

Long Range Planning Session

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# Nuclear isotopes are used in many household items and are pushing the scientific frontier in applications

- Household items
- Industry
- Scientific research



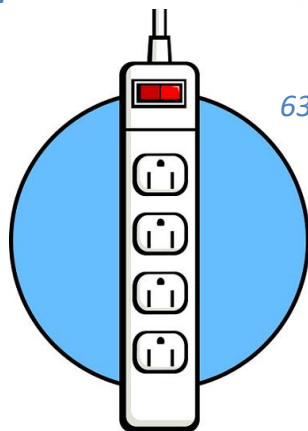
$^{85}\text{Kr}$ : indicator lights on household appliances



$^{241}\text{Am}$ : fire/smoke alarm



$^{238}\text{U}$ ,  $^{nat}\text{U}$ ,  $^{40}\text{K}$ : Dentures and crowns (<1980's)



$^{63}\text{Ni}$ : surge protector



$^{147}\text{Pm}$ : electric blanket thermostats

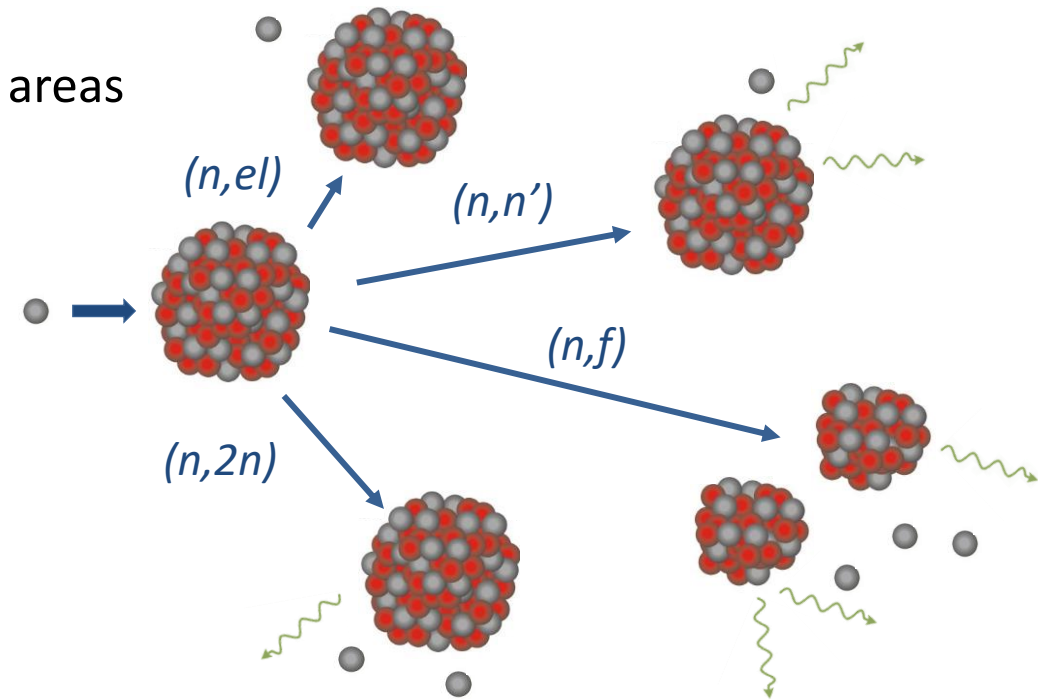


$^{229}\text{Th}$ : Helps fluorescent lights last longer

Criticality Safety  
Health Physics  
Isotope Biogeochemistry  
Isotope Geochemistry  
Isotope Hydrology  
Nuclear Astrophysics  
Nuclear Chemistry  
Nuclear Engineering  
Nuclear Geophysics  
Nuclear Medicine  
Nuclear Physics  
Radiobiology  
Radiochemistry  
Radioecology

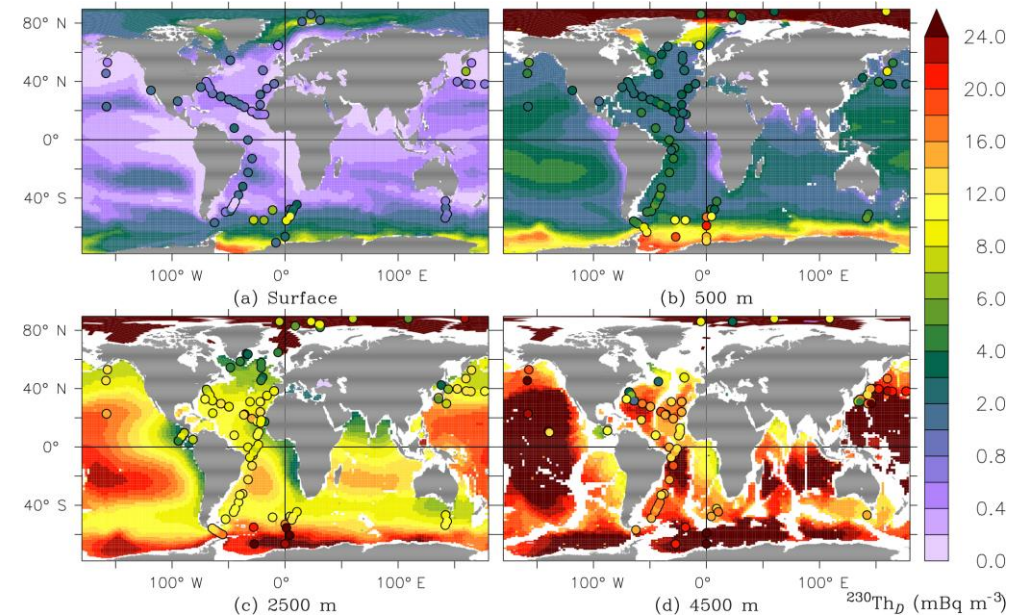
# Today will touch on uses, opportunities, and challenges

- Nuclear data are used in many different areas benefitting society – they won't all be covered here, this will be a whirlwind overview
- Some examples will be given, but are not all-encompassing
- Utilization of nuclear data can be generally grouped into three areas
  - Chronometers
  - Tracers/signatures
  - Energy deposition and/or production
- Probability for the event and what is produced
  - Radioactive decay half-life, reaction cross section
  - Emissions and products
    - Types
    - Number, multiplicities
    - Energies
    - Angles



# Understanding (and improving) our natural world

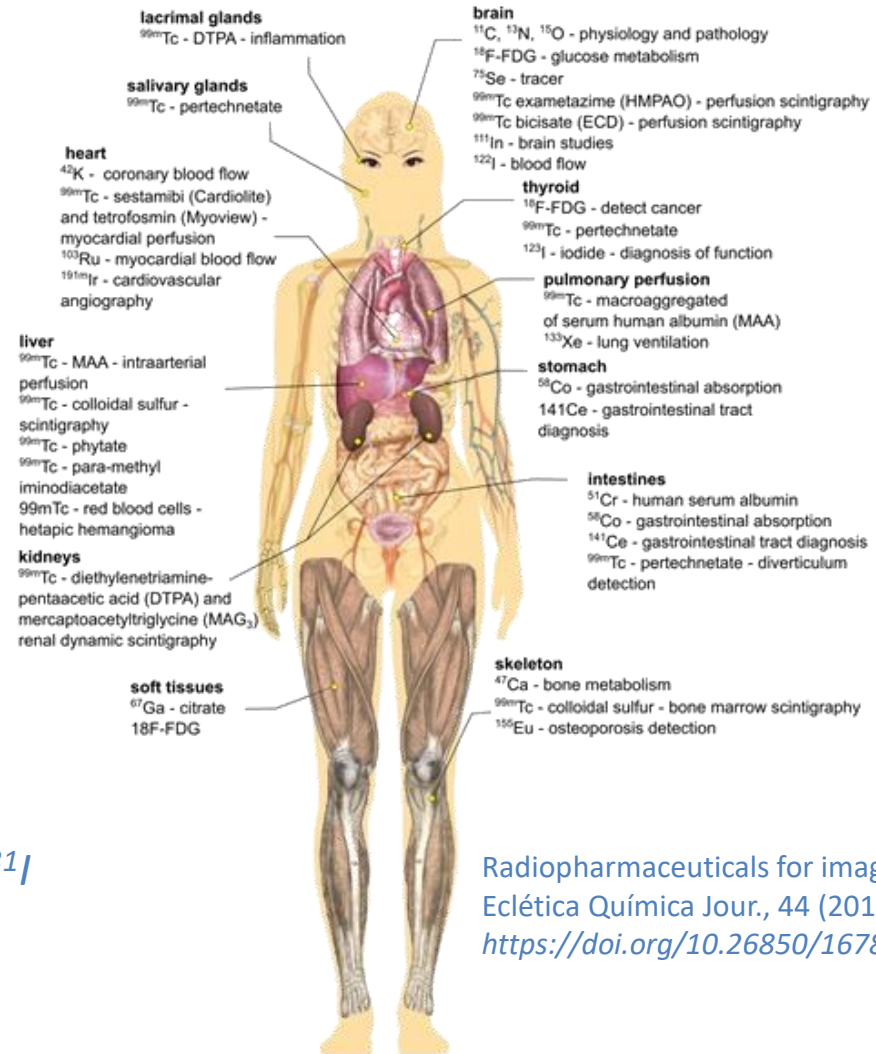
- Astrophysics
  - Nucleosynthesis and stellar energy generation; *Reactions ( $cp$ ,  $n$ ,  $\gamma$ ) and decays for stable and unstable*
  - Chronometers/Dating;  $^{187}\text{Re}$ ,  $^{232}\text{Th}$ ,  $^{235}\text{U}$ ,  $^{238}\text{U}$ ...  $^{40}\text{K}$ ,  $^{87}\text{Rb}$ ,  $^{138}\text{La}$ ,  $^{147}\text{Sm}$ ,  $^{176}\text{Lu}$
- Oceanography
  - Food chain contamination;  $\text{Co}$ ,  $\text{Ag}$ ,  $\text{Zn}$ ,  $\text{Cd}$ ,  $\text{Hg}$ ,  $^{137}\text{Cs}$ ,  $^{210}\text{Po}$ ,  $^{210}\text{Pb}$ ,  $^{241}\text{Am}$
  - Ocean acidification;  $^{13,14}\text{C}$
  - Currents and mixing;  $^3\text{H}$ ,  $^{14}\text{C}$ ,  $^{90}\text{Sr}$ ,  $^{129}\text{I}$ ,  $^{137}\text{Cs}$ ,  $^{238,239,240}\text{Pu}$ ,  $^{241}\text{Am}$
- Climate modeling
  - Ice caps/cores;  $^{18}\text{O}/^{16}\text{O}$ ,  $^2\text{H}/^1\text{H}$ ,  $^{10}\text{Be}$ ,  $^{36}\text{Cl}$
  - Tree rings;  $^{13,14}\text{C}$ ,  $^{226,228}\text{Ra}$
- Agriculture
  - Fertilization;  $^{15}\text{N}$ ,  $^{32}\text{P}$
  - Radiation-induced mutations;  $^{60}\text{Co}$ ,  $\beta$ ,  $cp$ ,  $n$ , *cosmic*
  - Irradiation/food safety;  $^{60}\text{Co}$ ,  $^{137}\text{Cs}$



Model of dissolved  $^{230}\text{Th}$  activity; Hulten *et al.*, *Geosci. Model Dev.*, 11, (2018) pp 3537–3556  
<https://doi.org/10.5194/gmd-11-3537-2018>

# Understanding (and improving) our human bodies

- Irradiation
  - Cancer treatment
    - External beam radiotherapy;  $^{60}\text{Co}$ ,  $^{137}\text{Cs}$ ,  $\beta$ ,  $cp$ ,  $n$
    - Brachytherapy;  $^{125}\text{I}$ ,  $^{131}\text{I}$ ,  $^{192}\text{Ir}$
  - Allograft sterilization;  $^{60}\text{Co}$ ,  $^{137}\text{Cs}$
- Diagnostic
  - imaging, PET, SPECT;  $^{18}\text{F}$ ,  $^{68}\text{Ga}$ ,  $^{82}\text{Rb}$ ,  $^{99\text{m}}\text{Tc}$ ,  $^{201}\text{Tl}$
- Therapeutic
  - Pain relief;  $^{89}\text{Sr}$ ,  $^{153}\text{Sm}$ ,  $^{186}\text{Re}$ ,  $^{169}\text{Er}$
  - Targeted energy deposition;  $^{90}\text{Y}$ ,  $^{131}\text{I}$ ,  $^{177}\text{Lu}$ ,  $^{225}\text{Ac}$ ,  $^{223}\text{Ra}$
- Theragnostic
  - Diagnostic and therapeutic;  $^{64}\text{Cu}$ - $^{67}\text{Cu}$ ,  $^{43,44}\text{Sc}$ - $^{47}\text{Sc}$ ,  $^{123}\text{I}$ - $^{131}\text{I}$
- Historical pacemaker,  $^{238}\text{Pu}$



Radiopharmaceuticals for imaging; Payolla *et al.*, *Eclética Química Jour.*, 44 (2019) pp 11–19  
<https://doi.org/10.26850/1678-4618eqj.v44.3.2019.p11-19>

# Reaching out (or in) to the unknown

- Radioisotope thermionic power;  $^{90}\text{Sr}$ ,  $^{238}\text{Pu}$ ,  $^{241}\text{Am}$ ,  $^{244}\text{Cm}$ 
  - Heat from radioactive decay
  - Arctic, ocean floor, space...
- Terrestrial exploration
  - Geochronometers;  $^{235,238}\text{U} > ^{207,206}\text{Pb}$ ,  $^{187}\text{Re}$ - $^{187}\text{Os}$ ,  $^{40}\text{Ar}/^{39}\text{Ar}$
  - Moon formation;  $^{182}\text{Hf}$ - $^{182}\text{W}$ ,  $^{146}\text{Sm}$ - $^{142}\text{Nd}$ , and  $^{176}\text{Lu}$ - $^{176}\text{Hf}$
  - Isotope hydrology, water age/origins;  $^3\text{H}$ ,  $^{14}\text{C}$ ,  $^{81}\text{Kr}$ ,  $^4\text{He}$
  - Geologic characterization; *NAA, PGAA, XRF, PIXE, PIGE, NMR*
- Extraterrestrial and meteoritic composition
  - Chondrites and chondrules; *U/Pb ratios,  $^{26}\text{Al}$ - $^{26}\text{Mg}$*
  - Iron meteorites;  $^{182}\text{Hf}$ - $^{182}\text{W}$
  - Achondrites, angrites;  $^{53}\text{Mn}/^{55}\text{Mn}$
  - Characterization
    - asteroid Psyche;  *$\gamma$ -spectroscopy*
    - Saturn moon Titan; *neutron activation,  $\gamma$ -spectroscopy*

TABLE I. Physical properties, abundance and occurrence of the geologically important radioactive elements.

Element	Active isotope	Radiations	Energy in Mev	Decay constant $\lambda$ in $\text{sec}^{-1}$	Half-period $T$	‡Mean abundance of active isotope g/g rock	Occurrence
Potassium	$\text{K}^{40}$	$\beta$	1.2	$1.4 \pm 0.3 \times 10^{-17}$	$1.6 \pm 0.3 \times 10^9 \text{y}$	$2.2 \pm 0.4 \times 10^{-6}$	Rocks, evaporites, and sea-water.
		$\gamma$	2.0	$4 \times 10^{-19}$	$5 \times 10^{10} \text{y}$		
Rubidium	$\text{Rb}^{87}$	$\beta$	0.13	$3.5 \pm 1 \times 10^{-19}$	$6.3 \pm 2 \times 10^{10} \text{y}$	ca $6 \times 10^{-6}$	Lepidolite and other Li-bearing minerals.
Samarium	$\text{Sm}^{148}$	$\alpha$	2.55	$2.5 \times 10^{-19}$	$0.9 \times 10^{11} \text{y}$	ca $10^{-6}$	Samarskite, cerite and monazite.
Radon	$\text{Rn}$	$\alpha$	5.49	$2.10 \times 10^{-6}$	3.82d	ca $3 \times 10^{-18}$ ca $7 \times 10^{-22}$ ca $2 \times 10^{-24}$	Genetically associated with uranium, thorium, and actino-uranium, respectively, in rocks, radioactive minerals, atmosphere, mineral waters, and sea water.
		$\alpha$	6.28	$1.27 \times 10^{-2}$	54.5s		
		$\alpha$	6.82	0.177	3.92s		
Radium	$\text{Ra}$	$\alpha$	4.87	$1.4 \times 10^{-11}$	1600y	$0.5 \pm 0.1 \times 10^{-12}$ ca $4 \times 10^{-18}$	Genetically associated with uranium and thorium, respectively, also found separately in certain mineral springs.
		$\alpha$	5.78	$2.20 \times 10^{-6}$	3.64d		
Thorium	$\text{Th}$	$\alpha$	4.31	$1.58 \times 10^{-18}$	$1.40 \times 10^{10} \text{y}$	$5.6 \pm 1 \times 10^{-6}$ ca $8 \times 10^{-16}$ ca $3 \times 10^{-11}$	Rocks, monazite, thorite, and thorianite. Ionium genetically associated with uranium.
		$\alpha$	5.47	$1.16 \times 10^{-8}$	1.9y		
		$\alpha$	4.66	$3 \times 10^{-13}$	$8 \times 10^4 \text{y}$		
Protactinium	$\text{Pa}$	$\alpha$	5.1	$7.0 \times 10^{-13}$	$3.1 \times 10^4 \text{y}$	ca $4 \times 10^{-13}$	Genetically associated with actino-uranium ( $\text{U}^{238}$ ).
Uranium	$\text{U}^{238}$	$\alpha$	4.13	$4.87 \times 10^{-18}$	$4.52 \times 10^8 \text{y}$	$1.4 \pm 0.3 \times 10^{-6}$ $1.0 \times 10^{-8}$	Rocks, uraninite pitchblende and carnotite.
		$\alpha$	4.4	$3.1 \pm 0.1 \times 10^{-17}$	$7.1 \pm 0.3 \times 10^8 \text{y}$		

Nuclear geophysics is a mature field where the use of isotopes was recognized early; Goodman, J. of Appl. Physics 12 (1942) pp 276 - 289  
<https://doi.org/10.1063/1.1714866>

# Industrial applications utilize decays and reactions

- Nuclear densitometry;  $^{137}\text{Cs}$ ,  $^{226}\text{Ra}$ ,  $^{241}\text{Am/Be}$ 
  - Nuclear logging, Measurement While Drilling, Logging While Drilling
  - Identify oil deposits, porosity, water content of coal
  - Construction, soil compaction and asphalt
- Oil and gas tracing;  $^{24}\text{Na}$ ,  $^{56}\text{Mn}$ ,  $^{110\text{m}}\text{Ag}$ ,  $^{131}\text{I}$ ,  $^{133}\text{Xe}$ ,  $^{140}\text{La}$
- Processing;  $^{82}\text{Br}$ ,  $^{137}\text{Cs}$ ,  $^{109}\text{Cd}$ ,  $^{192}\text{Ir}$ ,  $^{244}\text{Cm}$ 
  - Flow, blending, separation of materials
  - Automatic control of filling furnaces, vessels, transport containers
- Materials characterization
  - Construction, e.g. welds, wear, damage effects, shielding;  $^{137}\text{Cs}$ ,  $^{60}\text{Co}$ ,  $^{192}\text{Ir}$
  - Contamination, mining, processing, and smelting of uranium, phosphate, lead and iron ore, burning fossil fuels (coal), and burning leaded gasoline;  $^{238}\text{U}$  and  $^{232}\text{Th}$  decay series ( $^{210}\text{Pb}$ ,  $^{210}\text{Po}$ ),  $\text{Rn}$  gas
- Transmutation, e.g. doping Si semiconductors; *reactor n (fission)*



Paper/board weight using  $^{85}\text{Kr}$  or  $^{147}\text{Pm}$  (Beta-emitters)  
<https://jasch.net.in/paper-global-trendsetting-technology>

# Nuclear reactors provide power

- Electrical generation
  - 437 operating power reactors, 57 under construction, 102 planned (Sept 2022); mostly Pressurized Water Reactor (PWR)
  - Higher burn-up, modified fuels proposed
  
- Propulsion
  - aircraft carriers, nuclear submarines, ice breakers
  - space
    - Nuclear Thermal Propulsion (NTP)
    - Nuclear Electric Propulsion (NEP)
  
- Future Reactors
  - Small Modular Reactors, 50+ concepts;
    - fast neutron reactions, materials, dynamics*
  - Fusion Reactors; *cp fusion*
  
- Safeguards
  - Material control and accounting
  - Characterization

*Fission, fission product decays and reactions, neutron absorption and moderation reactions, n and  $\gamma$  production and transmission/shielding, half-lives...*

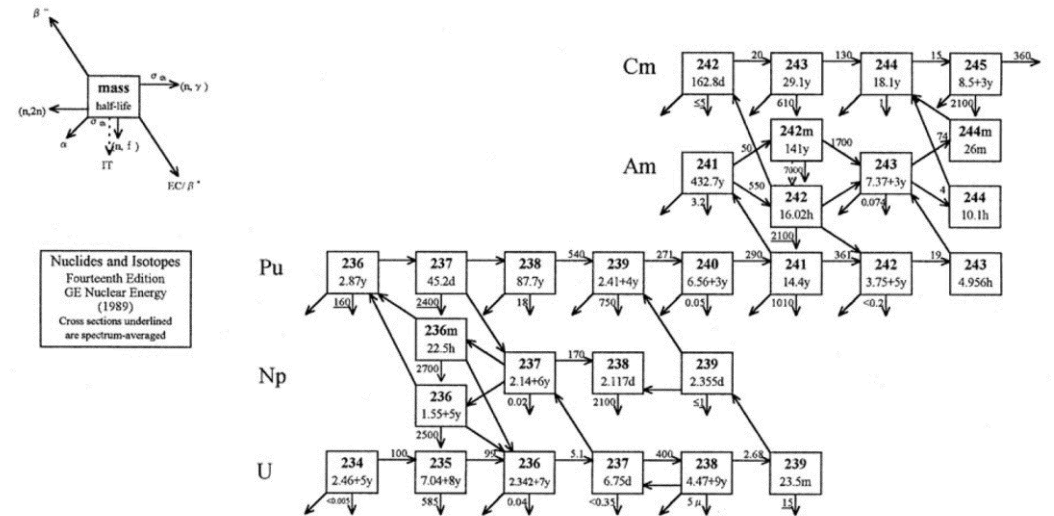


FIG.III-3. Actinide transmutation chains for  $UO_2$  and MOX fuel (detail).

A large number of complex reaction networks in the fuel and other materials must be considered; Nuclear Fuel Cycle Simulation System: Improvements and Applications, IAEA-TECDOC-1864



# Nuclear security considers nuclear threats

- Non-proliferation: prevent the spread of nuclear weapons
  - Detect, characterize, monitor nuclear fuel cycle activities
  - Detect the illicit movement of nuclear materials
- Counterterrorism and counterproliferation
  - Counter acquisition of nuclear capabilities
  - Respond to nuclear and radiological incidents
- Forensics
  - Determine origin of materials outside of regulatory control
  - Respond to nuclear and radiological incidents
- Stockpile stewardship
  - Expand predictive capability
  - Maintain deterrence

*Fission, fission product decays and reactions, neutron reactions,  $\alpha$  reactions,  $n$  and  $\gamma$  production and transmission/shielding, half-lives...*



Radiation portal monitor  
[https://en.wikipedia.org/wiki/Radiation\\_portal\\_monitor](https://en.wikipedia.org/wiki/Radiation_portal_monitor)

“Cop with scary gadget”, blog of Zack Hample, 2008  
<https://mlblogsnaggingbaseballs.wordpress.com/2008/07/17/2008-home-run-derby>

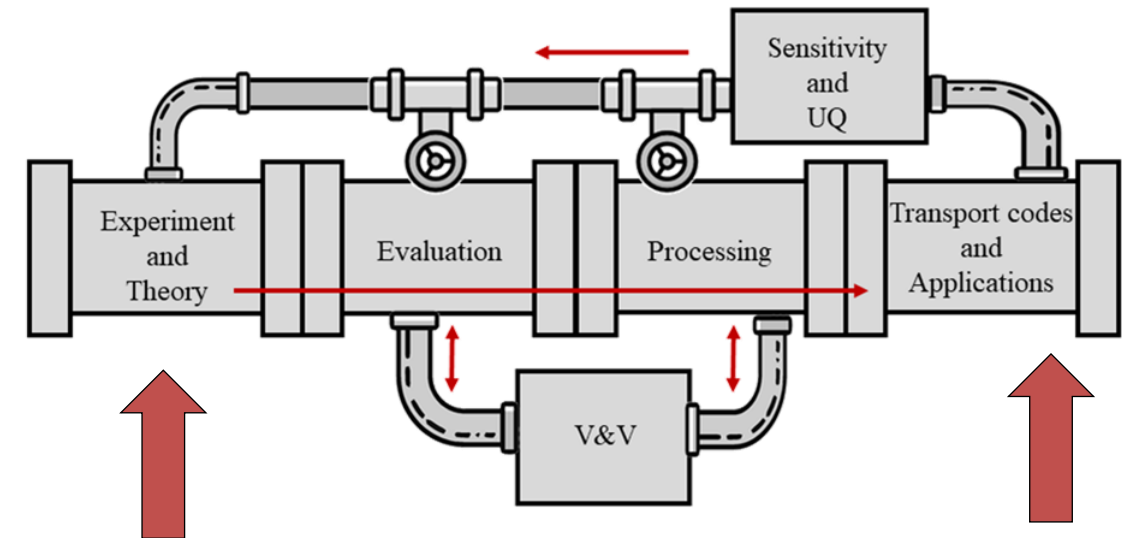
# Opportunities and challenges in nuclear applications

- The use of nuclear reactions and radioisotopes has many significant advantages
  - Security can be gained: nuclear, energy, food, and water
  - Scientific understanding of complex processes can be undertaken - fact-based findings can support policymakers, e.g. pollution control, public safety
  - Information and controls can provide significant cost and time savings, as well as safety
- Most applications are multidisciplinary
  - Utilization of nuclear science techniques often requires training and knowledge
  - What can we do to support the next generation in connecting fields?
- Risk
  - Regulations and controls can inhibit use of radioactive materials
  - What can we do to properly understand and communicate the risk/reward balance?
- Most applications rely on simulation and modeling
  - Do we have the necessary data readily available? *I think we can do better!*

# New data are needed, with improved timelines for delivering to applications

- Experiments and Theory are pushing the scientific boundaries
  - New experimental facilities, complex detector arrays
  - Sophisticated algorithms, analyses
  - Larger computation platforms, GPU capabilities
- Data needs
  - Reactions on unstable isotopes
  - Quantified uncertainties; precision improvements may need integral or quasi-integral data
  - Nuclear structure and decay data with reaction data
- Evaluation capabilities
  - Rate-limiting step
  - Modern tools and infrastructure
  - Peoplepower

*Nuclear data pipeline*



**Foundational Sciences**

**Applications**

*Nuclear data are shared: different applications use the same data resource*

# Multiple applications rely on nuclear data

- Six examples of applications fields
  - Environmental sciences
  - Medical physics
  - Geophysical characterization
  - Industrial applications
  - Nuclear power
  - Nuclear security
- Multidisciplinary
- Difficult environments –
  - Underwater, underground, outer space
  - Extreme temperatures, pressures
- Challenging observations/detection... and interpretation

*The nuclear science community “owns” part of the problem space:*

- Can we produce new tracer or medical isotopes?  
Can we calculate the production accurately?
- Do we understand the nuclear reaction networks in high flux environments?
- Do we know the reaction and decay products accurately?
- Are these data accessible for modeling/simulation?
- Do we have detection techniques that could be applied in the field?



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# Naturally occurring isotopes from U and Th series

Element	U-238 series						Th-232 series				U-235 series			
Neptunium														
Uranium	U-238 $4.47 \times 10^9$ y		U-234 $2.48 \times 10^5$ y								U-235 $7.04 \times 10^8$ y			
Protactinium		Pa-234 1.18 min										Pa-231 $3.25 \times 10^4$ y		
Thorium	Th-234 24.1 d		Th-230 $7.52 \times 10^4$ y				Th-232 $1.40 \times 10^{10}$ y	Th-228 1.91 y			Th-231 25.5 hrs		Th-227 18.7 d	
Actinium								Ac-228 6.13 hrs				Ac-227 21.8 y		
Radium			Ra-226 $1.62 \times 10^3$ y					Ra-228 5.75 y	Ra-224 3.66 d				Ra-223 11.4 d	
Francium														
Radon			Rn-222 3.82 d						Rn-220 55.6 s				Rn-219 3.96 s	
Astatine														
Polonium			Po-218 3.05 min	Po-214 $1.64 \times 10^{-4}$ s	Po-210 138 d			Po-216 0.15 s	64%	Po-212 $3.0 \times 10^{-7}$ s			Po-215 $1.78 \times 10^{-3}$ s	
Bismuth				Bi-214 19.7 min	Bi-210 5.01 d					Bi-212 60.6 min				Bi-211 2.15 min
Lead			Pb-214 26.8 min	Pb-210 22.3 y	Pb-206 Stable lead isotope			Pb-212 10.6 hrs	36%	Pb-208 Stable lead isotope			Pb-211 36.1 min	Pb-207 Stable lead isotope
Thallium										Tl-208 3.05 min				Tl-207 4.77 min